



Properties of Silica-SBR Compounds Using Cellulose Dispersant Applicable to Tire Tread Rubber

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Abstract: Silica-styrene butadiene rubber (Silica-SBR) compounds have been used in the preparation of tire treads. The silica dispersibility of silica-SBR compounds is related to the processability, mechanical properties, and wear resistance of tires. Recently, in order to improve the silica dispersibility of the silica-SBR compounds, the wet masterbatch (WMB) process was introduced, which is a method of mixing rubber in the water phase. We aimed to improve the silica dispersibility of the silica-SBR compounds by preparing a silica dispersant applicable to the WMB process. For this purpose, cellulose, 2-hydroxyethyl cellulose, and cellulose acetate were employed as a silica dispersant. The silica dispersibility of the compounds was measured by a moving die rheometer. Improvement in the processability of silica-SBR compounds was evaluated by the Mooney viscometer. The wear resistance of silica-SBR compounds using a cellulose dispersant was improved by up to 29%.

Keywords: Cellulose, Silica, SBR, Dispersant, Wet masterbatch

Introduction

Automotive tires are composed of ten or more layers, such as treads, sidewalls, composites, belt layers, and carcasses. Among them, the tread is the outermost layer of the tire, which is the most important layer because it contacts the ground directly. The tread is made of rubber that is resistant to breakage and impact, to protect the conductor and belt layers inside the tire, and wear-resistant rubber to increase tire life.

It is difficult to simultaneously improve the rolling resistance, wear resistance, and wetting property of the tire tread. Among these three properties, improvement of one physical property causes deterioration of another. Moreover, controlling these properties is very important for the manufacturing tire tread. The wear of the tire describes the phenomenon whereby the surface of the tread rubber contacting the ground is worn out because of the frictional force generated between the tread and the road surface. The wear of the tire considerably affects the life and braking performance of the tire. In addition, in terms of cost reduction, the tire tread should have excellent wear resistance.¹

Owing to their good dispersion characteristics, carbon black fillers can facilitate the preparation of good-quality rubber compounds. However, the hysteresis of the rubber compounds increases, which is disadvantageous in terms of the fuel economy of the tire.² In recent years, rubber compounds containing silica fillers have become increasingly popular because of reduced rolling resistance in tires and a significant improvement in the frictional force on wet and snow-covered roads. In particular, the silica content of ultra-high-performance tires and winter tires sold recently may exceed 100 phr. However, there are many difficulties associated with the preparation of compounded rubber using silica that exhibits satisfactory performance.^{3,4}

Since silanol groups are present on the surface of the silica, hydrogen bonds can form between the hydroxyl groups on the silica surface; thus, the silica dispersibility of the rubber compounds decreases, causing aggregation of the silica particles. The phenomenon of silica aggregation adversely affects the properties of the rubber compound.⁵ Recently, many studies have been developed to improve the dispersibility of composites in various materials fields, and the wet masterbatch (WMB) process was developed to improve the silica dispersibility of rubber compounds.⁶ WMB rubber compounds are prepared by obtaining a solid rubber com-

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pound using a flocculant to solidify the sludge in which the filler-containing dispersion is mixed with the rubber latex. The use of the WMB process improves silica dispersibility and thus the mechanical properties of the resulting rubber compounds, in comparison to that achieved by mixing rubber and silica through the dry masterbatch (DMB) process. Typical rubber compounds manufactured by the WMB process have low rolling resistance and improved fatigue resistance.⁷

Previous studies have shown that the WMB process can load more than 100 phr of silica into the rubber compound; however, this reduces silica dispersion, which adversely affects the mechanical properties, abrasion resistance, and rolling resistance of the compound. While various silica dispersants are commercially available as additives for compounds in the DMB process, commercial silica dispersants suitable for the WMB process have not yet been explored.⁸⁻¹⁰ Recently, the development of silica dispersants applicable to WMB process has gained considerable attention. Some researchers have prepared dispersants using anionic polymers such as poly (acrylic acid), styrene-acrylic acid copolymers, and vinyl naphthalene-acrylic acid copolymers to enhance filler dispersion in WMB.^{11,12}

Currently, researchers are focusing on the development of eco-friendly tire materials as a measure to reduce global warming; thus, rubber additives such as silica dispersants must be developed using eco-friendly materials.^{13,14} An example of relevant and eco-friendly material is cellulose, which is naturally available and has the chemical composition similar to that of starch. Previous studies have shown that the addition of cellulose to rubber compounds in the DMB process improved the environmental friendliness, biodegradability and processability, mechanical properties, and dynamic mechanical performance of the tire.^{15,16} Since cellulose is composed of carbon chains and hydroxyl groups, it exhibits both hydrophilic and hydrophobic functionalities. Because of these characteristics, research has been conducted using cellulose as a dispersant in various fields.¹⁷⁻¹⁹ However, the use of cellulose as a silica dispersant for tire manufacturing using the WMB process has not yet been investigated.

In this study, cellulose variants having large numbers of hydroxyl groups were investigated for their use as silica dispersants in the WMB process with the aim of preventing aggregation and improving the silica dispersibility. In the WMB process, rubber compounds were prepared using various cellulosic materials as silica dispersants. In order to assess the silica dispersibility, the processability, mechanical

properties, and wear resistance of the rubber compounds were evaluated.

Experimental

1. Materials

Cellulose (Powder 20 μm , Sigma Aldrich, United States), 2-hydroxyethyl cellulose (Powder, Sigma Aldrich, United States), and cellulose acetate (Powder, Sigma Aldrich, United States) were used without further purification. ESR 1712 (KUMHO PETROCHEMICAL, South Korea) was the rubber latex used for the WMB process, and a 2% w/w CaCl_2 aqueous solution was used as a coagulant. Modified silica NK153 (MIRAESI, South Korea) was used as a reinforcing agent for the compound, whose surface was hydrophobically modified with 10% w/w bis [3-(triethoxysilyl) propyl] tetrasulfide (TESPT). Various compounding processing additives such as zinc oxide (ZnO), stearic acid (S/A), and N-(1,3-dimethyl-butyl)-N'-phenyl-p-phenylenediamine (6PPD, Sigma Aldrich, United States) were used. In the final compounding step, sulfur (SAMCHUN, South Korea) was used as a crosslinking agent. N-cyclohexyl-2-benzothiazole sulfonamide (CBS, Tokyo Chemical Industry, Japan) and diphenyl guanidine (DPG, Sigma Aldrich, United States) were used as crosslinking accelerators.

2. Preparation of the silica-SBR compounds

The silica-SBR compounds were prepared using three processes: WMB, silica master match (SMB), and final master batch (FMB). In the WMB process, the emulsion styrene butadiene rubber (ESBR) was added to silica sludge. After stirring, the cellulose-based dispersants were added, as solutions dissolved in water, and the coagulant was added while stirring to aggregate the ESBR and load the silica. In the SMB process, ZnO, S/A, and 6PPD were added as the primary mixing agents. The FMB process is a secondary mixing process that prepares the silica-SBR compounds before vulcanization by adding sulfur for vulcanization as well as DPG and CBS as vulcanization accelerators.

Figure 1 illustrates the WMB, SMB and FMB processes for the preparation of silica-SBR compounds. TESPT-modified silica (2% w/w) was added to distilled water and stirred at 60°C to prepare a silica slurry. After mixing it with the ESBR, also heated to 60°C, the mixture was stirred for 10

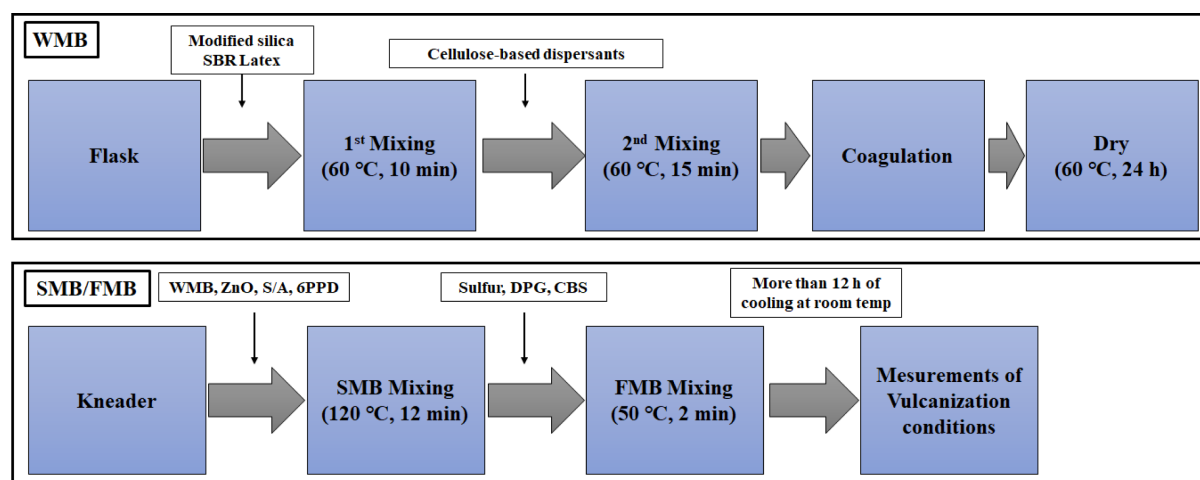


Figure 1. WMB, SMB and FMB process for the preparation of a silica-SBR composite.

min. A solution prepared by dissolving the cellulose-based dispersants in distilled water was then added to the silica and ESBR slurry. After 15 min the mixed slurry was coagulated with CaCl_2 , washed once, and dried at 60°C for 24 h.

The SMB was produced using dried WMB compounds. The dried WMB compounds were mixed with ZnO , S/A, and 6PPD for 12 min at 120°C . The FMB was prepared using the SMB compounds, CBS, DPG, and sulfur. They were mixed for 2 min at 50°C using a two-roll mill for the preparation of the final silica-SBR compounds. The detailed formulations of silica-SBR compounds are shown in Table 1.

The cure characteristics and silica dispersibility of four silica-SBR compounds made according to Table 2 were measured by MDR. The MDR results confirmed that silica dispersibility was improved when cellulose was added for all three types of cellulose-based dispersants. Next, silica-SBR compounds were prepared with cellulose applied at 2, 4, and 6 phr levels, and the content of cellulose was confirmed.

The torque and cure time of the silica-SBR compounds in the FMB stage were measured with a moving die rheometer (MDR, MDR2020, MYUNGJI TECH) at $160 \pm 1^\circ\text{C}$ for 30

Table 2. Cure Characteristics Data of the Silica-Rubber Compounds by Cellulose-Based Dispersants

	Unit	WN	WC2	WHC2	WCA2
t_{10}	min:ss	03:15	03:44	03:54	03:45
t_{90}	min:ss	15:51	15:33	15:16	14:41
T_{\min}	N-m	0.561	0.545	0.583	0.554
T_{\max}	N-m	2.011	1.885	1.874	1.799
$T_{\max}-T_{\min}$	N-m	1.450	1.340	1.291	1.245

min. The optimum cure time was determined using an MDR based on ISO 6502. This test measures the minimum torque value (T_{\min}), maximum torque value (T_{\max}), scorch time (t_{10}), and optimum cure time (t_{90}).

3. Mooney viscosity

The Mooney viscosity (MV, MV2020, MYUNGJI TECH) of the silica-SBR compounds in the FMB stage provides a measure of the processability of the unvulcanized rubber. In ML(1+4), the “M” means Mooney, the “L” means plate size, the “1” means 1 min preheating and the “4” corresponds to

Table 1. Formulation of WMB, SMB, and FMB for the Silica-SBR Compounds (phr)

	NAME	WN	WC2	WHC2	WCA2	WC4	WC6
WMB	E-SBR/Silica			100/100			
	TESPT			10			
	Cellulose	-	2	-	-	4	6
	2-hydroxyethyl cellulose	-	-	2	-	-	-
	Cellulose acetate	-	-	-	2	-	-
SMB	$\text{ZnO}/(\text{S/A})/6\text{PPD}$			3/(2)/1			
FMB	Sulfur/CBS/DPG			1.5/1.5/1.5			

the measured value after 4 min of rotor operation. Generally, when the value of ML(1+4) is low, the fluidity is considered to be good, and the processability to be excellent. The measurement of the MV (ML(1+4)@100°C) was performed using a Mooney viscometer as per ISO 289-1.

4. Mechanical properties and wear resistance

The modulus, tensile strength, and elongation of the silica-SBR compounds were measured according to the ASTM D412 standard. A universal testing machine (H5KT-0635, Tinius Olsen) was used to measure these parameters. The wear resistance of the silica-SBR compounds was measured using a DIN wear tester (AT-2020-D, MYUNGJI TECH) in accordance with the ASTM D 5963 standard.

Results and Discussion

1. Expected mechanism for the silica dispersing effect of cellulose

Figure 2 illustrates the dispersing mechanism of cellulose in the WMB process. In the case of modified silica, the TESPT-modified portion is hydrophobic, but the unmodified portion has hydroxyl groups. Silica aggregation occurs due to hydrogen bonding between the hydroxyl groups, which reduces the silica dispersibility of the silica-SBR compounds. Cellulose adheres to hydroxyl groups present on the

surface of silica via hydrogen bonding to prevent aggregation of silica, improving the silica dispersibility in the WMB process.

2. Cure characteristics

The cure properties of the four rubber compounds prepared according to the formulations of Table 1 were confirmed by MDR. The MDR results of silica-SBR compounds with and without cellulose-based dispersants are shown in Figure 3 and Table 2. The t_{10} indicates the time when the vulcanization of the silica-SBR compound is 10%, also known as the scorch time of the rubber. A longer t_{10} time correlates to better processability of the silica-SBR compound. Our results show that the t_{10} increased for both samples with cellulose-based dispersants. Specifically, for WHC, the t_{10} value increased by 40 s when compared to that of the corresponding material without a dispersant (WN). The t_{90} parameter represents the cure time when rubber vulcanization is 90%

Table 3. Cure Characteristics Data of the Silica-Rubber Compounds according to Cellulose Content

	Unit	WN	WC2	WC4	WC6
t_{10}	min:ss	03:15	03:44	03:37	03:41
t_{90}	min:ss	15:51	15:33	14:49	14:46
T_{min}	N-m	0.561	0.545	0.534	0.525
T_{max}	N-m	2.011	1.885	1.967	1.865
$T_{max}-T_{min}$	N-m	1.450	1.340	1.433	1.340

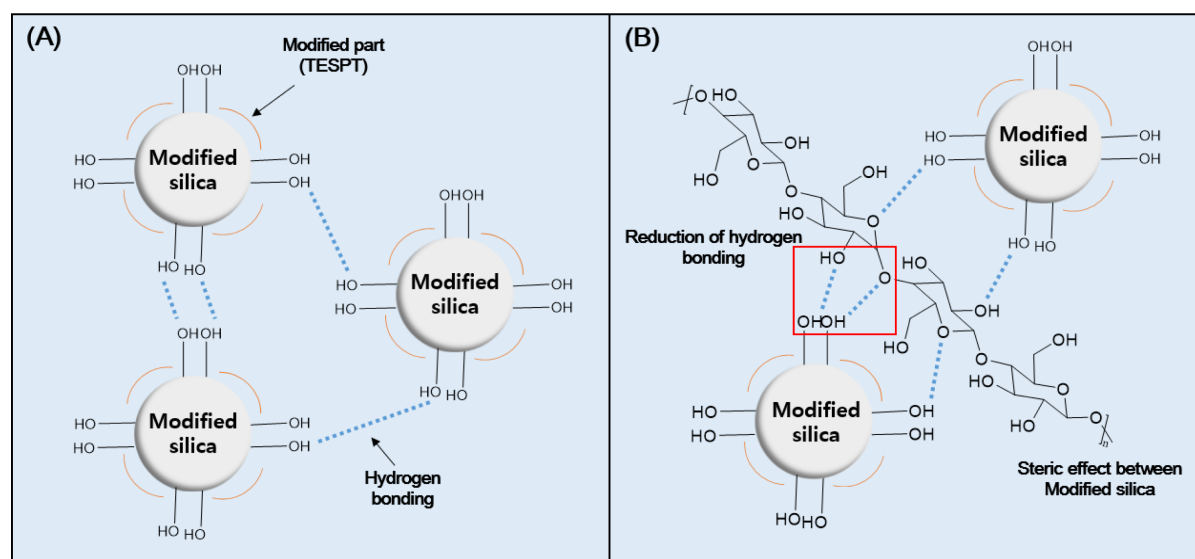


Figure 2. Illustration of expected filler network in the WMB process: (A) where a residual silanol group in the modified silica forms a hydrogen bond and (B) where a cellulose dispersant interacts with silanol group in the modified silica.

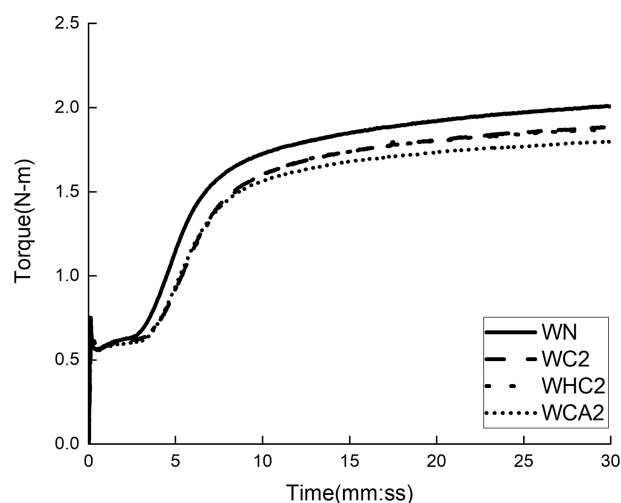


Figure 3. Cure characteristics of compounds by cellulose-based dispersants.

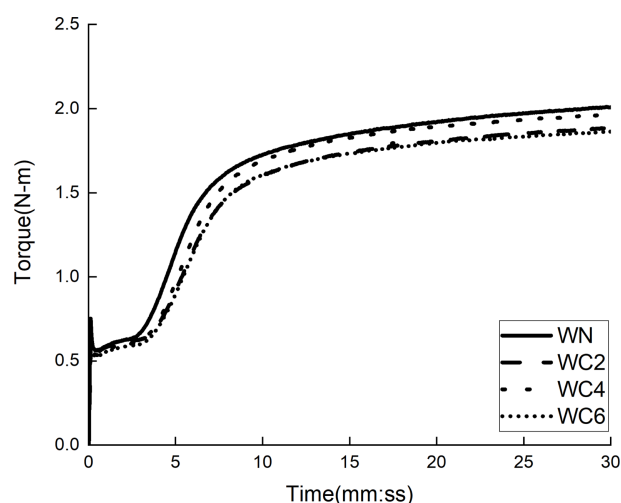


Figure 4. Cure characteristics of compounds according to cellulose content.

complete, and is called the optimum vulcanization time. In this study, the t_{90} values decreased for all three silica-SBR compounds prepared with cellulose-based dispersants. The T_{min} indicates the minimum torque of silica-SBR compounds and tends to decrease as the silica dispersibility improves. It has been shown that dispersants other than cellulose increased T_{min} when compared to rubber materials prepared without dispersants, which has an adverse effect on silica dispersibility. The T_{max} represents the maximum torque of the silica-SBR compound and tends to increase as the crosslinking density increases. In the case of cellulose-based dispersants, it was found that the T_{max} decreased, which resulted in a decrease in the crosslinking density of the rubber. Hydrogen bonding between the CBS cure accelerator and cellulose may have led to a loss of CBS activity. The $T_{max}-T_{min}$ represents the crosslinking density of the rubber compound, which is similar to the above-mentioned trend, and shows that the crosslinking density decreased when the cellulose-based dispersants were used. MDR results showed the lowest T_{min} and the highest crosslinking density for silica-rubber compounds prepared with cellulose (WC) among the three types cellulose-based dispersants used. Therefore, we chose to examine cellulose as a dispersant for the remainder of this study.

The cure characteristics of compounds according to cellulose content are shown in Figure 4 and listed in Table 3. As the cellulose content increases, the t_{90} decreases. In the case of WC06, the t_{90} was reduced by 1 min or more when compared with WN, which is due to the improved silica dispersibility. The T_{min} values measured, decreased with

increasing cellulose content, with WC06 being reduced by up to 7% when compared with WN. This shows that cellulose interferes with the aggregation of silica through hydrogen bonding with the silica. T_{max} and $T_{max}-T_{min}$ values of the WC series using cellulose dispersants decreased compared to those of WN, which was due to the lower activity of CBS caused by a higher degree of hydrogen bonding between CBS and cellulose.

3. MV measurements

Figure 5 and Table 4 provide comparisons of the MV values for the silica-SBR compounds in the FMB stage. ML(1+4) is the MV index, indicating the processability for the rubber material. A low MV value implies high process-

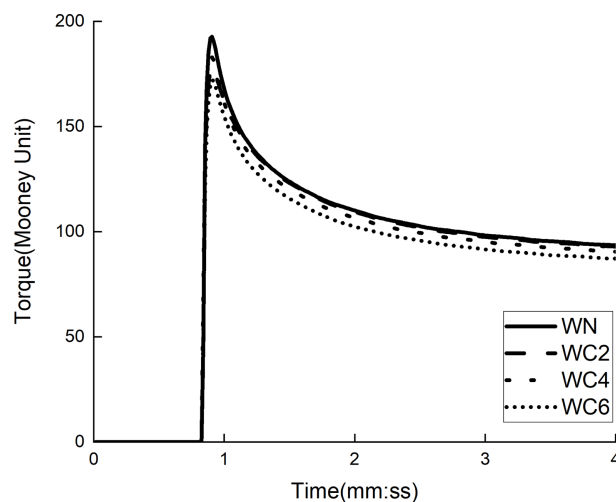


Figure 5. Mooney Viscosity Properties of compounds.

Table 4. MV Values of the Silica–Rubber Compounds

	Unit	WN	WC2	WHC2	WCA2	WC4	WC6
Initial	Torque	192.8	178.6	over	over	183.1	172.1
ML(1+4)	Torque	93.6	92.9	over	over	90.5	87.2

ability of the silica–rubber compound, indicating good silica dispersion.^{20,21} In case of the compounds of WHC and WCA, the MV index was impossible to measure, which imply that 2-hydroxyethyl cellulose and cellulose acetate are not suitable silica dispersants. As the cellulose content increased, the MV values of the silica–rubber compounds decreased. Similar to the T_{min} results of MDR, ML(1+4) decreased with increasing cellulose content by up to 7% for WC06 compared to WN. Cellulose has been shown to improve processability by preventing the agglomeration of silica and consequently improving the silica dispersibility of the silica–SBR compounds.

4. Mechanical properties

The mechanical properties of the silica–SBR compounds are listed in Table 5. The $M_{100\%}$ and $M_{300\%}$ represent the moduli of the silica–SBR compounds at 100% and 300% elongation, respectively. All mechanical properties of the WC compounds were higher than those of WHC and WCA compounds. As the content of cellulose increased from 0 to 4 phr of cellulose, the $M_{100\%}$ and $M_{300\%}$ values increased, and then decreased for WC06, which contains 6 phr of silica. This is because cellulose improves the silica dispersion in the silica–SBR compounds, thereby improving the tensile strength of the rubber, until the 4 phr mark. When the content of cellulose is higher, for example at 6 phr, the tensile strength of WC compounds is lowered because the remaining cellulose reduces the crosslinking density of the rubber by weakening the activity of the CBS vulcanization.

5. Wear resistance

Table 5. Mechanical Properties of the Silica–SBR Compounds

	Unit	WN	WC2	WHC2	WCA2	WC4	WC6
$M_{100\%}$	kg/cm ²	13.5	15.0	14.5	13.7	15.9	12.7
$M_{300\%}$	kg/cm ²	42.4	48.6	44.5	44.2	50.7	38.9
Elongation at break	%	1050	1050	850	820	980	1060
Tensile strength	kg/cm ²	245	250	205	200	262	236

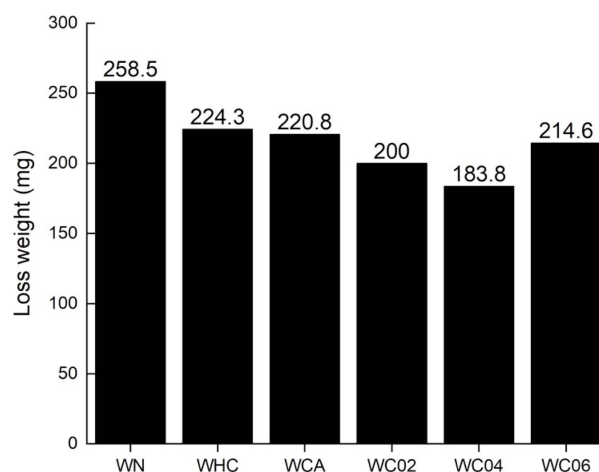
**Figure 6.** Wear resistance of the silica–SBR compounds.

Figure 6 graphically depicts the wear resistance of the silica–SBR compounds. The WN compound shows the highest weight loss among the compounds. Comparing the wear resistance of WHC and WCA compounds, those of WC series show the higher wear resistance, due to the poor silica dispersibility. Particularly, the WC04 compound had the best wear resistance with a weight loss decrease of up to 29% when compared to the rubber compound without cellulose (WN). This is believed to be due to improved silica dispersibility, resulting in smaller silica particles with increased surface area. Notably, the wear resistance deteriorated in the case of WC06. This is once again due to remaining cellulose in the compound resulting in lowered crosslinking density. This is also due to the deterioration of the activity of the vulcanization accelerator through hydrogen bonding between the cellulose and the vulcanization accelerator, similar to the tensile strength.

Conclusions

In this study, silica-SBR compounds were prepared by adding cellulose-based dispersants to the WMB process and subsequently evaluated. We confirmed that cellulose was the best dispersant through experiments comparing three cellulose-based dispersants, i.e., cellulose, 2-hydroxyethyl cellulose, and cellulose acetate. Vulcanization properties, processability, mechanical properties, and wear resistance of the cellulose containing compounds (WC) were measured. The cure rate of the WC series was higher than that of rubber compounds without cellulose. As the content of cellulose increased, the ML(1+4) values decreased, indicating that the silica dispersibility of the compounds had increased, thereby improving the processability of the silica-SBR compound. The mechanical properties of WC02 and WC04, with 2 and 4 phr of cellulose, respectively, were improved in terms of tensile strength. However, the mechanical properties of the WC compound with 6 phr were reduced. This seems to be a result of deactivation of CBS through hydrogen bonding between cellulose and CBS. The wear resistance of the WC compounds was enhanced with increasing cellulose content, but decreased again when cellulose content approached 6 phr. This is due to the presence of a large amount of cellulose having a low modulus, which is similar to the phenomenon observed for the tensile strength.

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References

1. B. J. Lee, K. W. Lim, S. C. Ji, K. Y. Jung, and T. J. Kim, "Advanced synthetic technology for high performance energy tire tread rubber", *Elastomers and Composites*, **44**, 232 (2009).
2. S. C. Han, S. J. Choe, and M. H. Han, "실리카 고무 배합 기술", *Rubber Technology*, **2**, 100 (2011).
3. G. Mathew, M. Y. Huh, J. M. Rhee, M. H. Lee, and C. Nah, "Improvement of properties of silica-filled styrene-butadiene rubber composites through plasma surface modification of silica", *Polymers for Advanced Technologies*, **15**, 400 (2004).
4. M. J. Wang, P. Zhang, and K. Mahmud, "Carbon—silica dual phase filler, a new generation reinforcing agent for rubber: Part IX. Application to truck tire tread compound", *Rubber Chemistry and Technology*, **74**, 124 (2001).
5. H. Eggers and P. Schummer, "Reinforcement mechanisms in carbon black and silica loaded rubber melts at low stresses", *Rubber Chemistry and Technology*, **69**, 253 (1996).
6. M. J. Kim, G. Sung, and J. Y. Sun, "Stretchable and reflective displays: materials, technologies and strategies", *Nano Convergence*, **6**, 1 (2019).
7. N. Minouchi and T. Miyasaka, "Rubber wet master batch, process for producing the same, rubber composition, and pneumatic tire", U.S. Patent No. 8,053,496. Washington, DC: U.S. Patent and Trademark Office (2011).
8. R. G. Seaman, "Rubber dispersing agent", U.S. Patent No. 1,867,633. Washington, DC: U.S. Patent and Trademark Office (1932).
9. J. Y. Lee, T. Lee, K. Kim, B. Kim, G. Kwag, J. Y. Kim, and H. J. Paik, "Poly (styrene-r-butadiene)-b-poly (poly (ethylene glycol) methyl ether methacrylate) as a silica dispersant in rubber compounds", *Polymer International*, **63**, 908 (2014).
10. I. Surya, H. Ismail, and A. R. Azura, "Alkanolamide as an accelerator, filler-dispersant and a plasticizer in silica-filled natural rubber compounds", *Polymer Testing*, **32**, 1313 (2013).
11. A. Nakayama, T. Kasai, M. Izuchi, and F. Nishiura, "Dispersion liquid of chemical for rubbers, method for producing same, rubber wet master batch containing chemical for rubbers, rubber composition, and tire", U.S. Patent Application No. 13/266,632 (2012).
12. F. Nishiura, Y. Kanda, and K. Someno, "Process for producing wet masterbatch, and wet masterbatch produced by said process", U.S. Patent No. 9,796,819. Washington, DC: U.S. Patent and Trademark Office (2017).
13. Q. Qi, Y. Wu, M. Tian, G. Liang, L. Zhang, and J. Ma, "Modification of starch for high performance elastomer", *Polymer*, **47**, 3896 (2006).
14. C. Liu, Y. Shao, and D. Jia, "Chemically modified starch reinforced natural rubber composites", *Polymer*, **49**, 2176 (2008).
15. K. Gopalan Nair and A. Dufresne, "Crab shell chitin whisker reinforced natural rubber nanocomposites. 1. Processing and swelling behavior", *Biomacromolecules*, **4**, 657 (2003).
16. W. Bai and K. Li, "Partial replacement of silica with microcrystalline cellulose in rubber composites", *Composites Part A: Applied Science and Manufacturing*, **40**, 1597 (2009).
17. N. Minami, Y. Kim, K. Miyashita, S. Kazaoui, and B. Nalini, "Cellulose derivatives as excellent dispersants for single-wall carbon nanotubes as demonstrated by absorption and photoluminescence spectroscopy", *Applied Physics Letters*, **88**, 093123 (2006).
18. Y. Li, H. Zhu, F. Shen, J. Wan, S. Lacey, Z. Fang, and L. Hu, "Nanocellulose as green dispersant for two-dimensional

- energy materials”, *Nano Energy*, **13**, 346 (2015).
19. E. Salahinejad, M. J. Hadianfard, D. D. Macdonald, M. Mozafari, D. Vashae, and L. Tayebi, “Zirconium titanate thin film prepared by an aqueous particulate sol–gel spin coating process using carboxymethyl cellulose as dispersant”, *Materials Letters*, **88**, 5 (2012).
20. S. Rooj, A. Das, K. W. Stöckelhuber, N. Mukhopadhyay, A. R. Bhattacharyya, D. Jehnichen, and G. Heinrich, “Pre-intercalation of long chain fatty acid in the interlayer space of layered silicates and preparation of montmorillonite/natural rubber nanocomposites”, *Applied Clay Science*, **67**, 50 (2012).
21. D. W. Kim, C. H. Kim, H. K. Jung, and Y. G. Kang, “The Effect of Glyceride Modified by Fatty Acid on Mechanical Properties of Silica filled Rubber Compounds”, *Elastomers and Composites*, **48**, 114 (2013).